

Nisbah Bobot Zeolit dan Arang Sekam Padi Sebagai Adsorben dalam Pengolahan Limbah Cair Laboratorium Kimia

Weight Ratio Of Zeolite and Rice Husk Charcoal As Adsorbant in the Chemical Laboratory's Wastewater Treatment

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Abstract

The purpose of this study is to determine the effectiveness of weight ratio of zeolite and rice husk charcoal in lowering the physico-chemical parameters of liquid waste chemical laboratory and determine the adsorption isotherm models weight ratio of zeolite and rice husk charcoal. Data were analyzed using Randomized Completely Block Design (RCBD), 5 treatments and 5 replications. As a block is the time analysis. As the treatment is different variations of the weight ratio of zeolite and rice husk (grams/gram), namely (5:5); (6:4); (7:3); (8:2) and (9:1). The treatment of its adsorbent were given separately and both of it is treated with stirring. To test the difference between treatment means, the Honestly Significant Differences (HSD) were used with 5% level of significance.

The results showed that the effectiveness of giving 8 grams zeolite for 240 minutes is 99.20% (COD); 68.78% (Cu) and 94.53% (Pb), respectively. While the effectiveness of giving 2 grams rice husk charcoal in the next 180 minutes are 42.34% (COD); 70.75% (Cu) and 84.30% (Pb), respectively. The maximum adsorption capacity of giving 8 grams zeolite within 240 minutes is $COD = 166.670 \text{ mg/g}$ (followed Freundlich isotherm model); $Cu = 0.327 \text{ mg/g}$ (followed Freundlich isotherm model); $Pb = 0.754 \text{ mg/g}$ (followed Langmuir isotherm model), respectively. While the adsorption capacity of giving 2 grams rice husk charcoal in the next 180 minutes are : $COD = 3.490 \times 10^5 \text{ mg/g}$ (followed Freundlich isotherm model); $Cu = 0.084 \text{ mg/g}$ (followed Freundlich isotherm model); $Pb = 0.782 \text{ mg/g}$ (followed Langmuir isotherm model), respectively.

Keywords: zeolite , rice husk charcoal, adsorbant, adsorption isotherms, waste water